

ABSOLUTE CONFIGURATION AND CONFORMATIONAL MOBILITY
OF DILOPHOL AND 3-ACETOXYACETYLDILOPHOL

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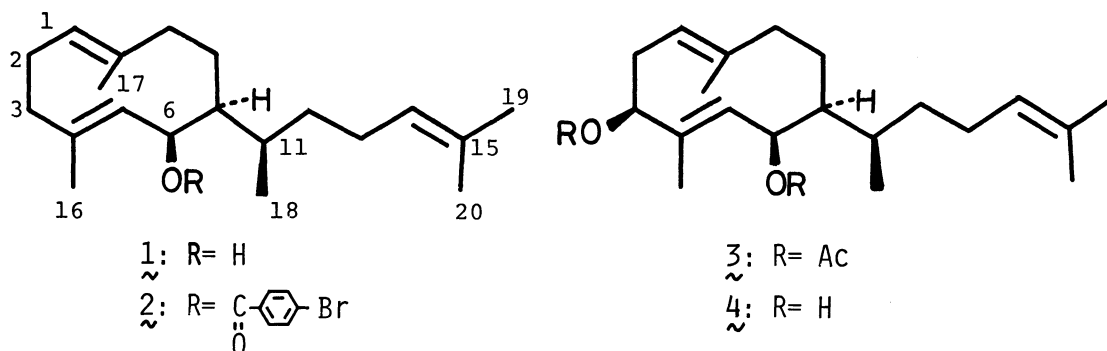
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Dilophol and 3-acetoxyacetyldilophol were isolated from *Dictyota dichotoma* and their absolute configurations and conformational behavior were clarified by X-ray analysis, NMR spectroscopy and molecular mechanics calculations.

Dilophol (**1**) is a potential precursor for many of the alga bicyclic diterpenes and was found in a Mediterranean alga *Dilophus ligulatus*.¹⁾ The structure of **1** was assigned as formula **1** based on NMR studies¹⁾ but absolute configuration remained unclear. We isolated **1** and its new analog, 3-acetoxyacetyldilophol (**3**), from a brown alga "Amijigusa" (*Dictyota dichotoma*) which was collected at Oshoro bay, Hokkaido. Determination of the absolute configurations and studies on the conformational aspects of the above two diterpenes, **1** and **3**, will be described below.

X-ray diffraction analysis was carried out on a crystalline p-bromobenzoate **2** (C₂₇H₃₇O₂Br, mp 84-85 °C). The crystal data were as follows: orthorhombic, space group p2₁2₁2₁, a=12.228(2), b=30.992(8), c=6.786(1) Å, Z=4, D_c=1.223 g cm⁻³. Intensity of 2376 unique reflections with 2θ values up to 130° were collected on a Rigaku four-circle diffractometer with graphite-monochromated Cu Kα radiation, using the ω-2θ scanning technique.²⁾ The structure was solved by the Monte Carlo direct method,³⁾ and was refined by the block-diagonal least-squares method with anisotropic thermal parameters. From the R ratio of 1.038,⁴⁾ the absolute configuration was established as shown in Fig. 1. Further least-squares refinements were performed including 34 hydrogen atoms; the final R value was 0.045.

The ¹H-NMR spectrum of **1** on a 60 MHz spectrometer showed peaks at δ 0.99 (3H, d, J=7 Hz, W_H of the single peak=2 Hz), 1.47 (3H, brs), 1.58 (6H, brs), 1.67 (3H,



brs), 4.57 (1H, brd, $J=8$ Hz), and 4.70–5.59 (3H, complex pattern) in CDCl_3 , while the spectrum on a 400 MHz instrument exhibited broad peaks at δ 1.02 (3H, $W_H=16$ Hz), 1.33 (6H, $W_H=16$ Hz), 1.61 (3H, brs), 1.70 (3H, brs), 4.57 (1H, $W_H=32$ Hz), 4.88 (1H, $W_H=44$ Hz), 4.99 (1H, $W_H=36$ Hz), and 5.28 (1H, t, $J=7$ Hz) in the same solvent. The ^{13}C -NMR spectrum (25 MHz) of **1** in CDCl_3 showed at room temperature merely broad signals due to carbon atoms of the ten-membered ring and sharp signals at δ 16.5, 17.4, 24.5, 25.7, 34.4, 124.3, and 131.6 due to those of the side chain. However, each of the broad signals observed at room temperature separated into a pair of sharp peaks with an intensity ratio of about 1:1 in toluene- d_8 solution at -28 °C. Moreover at higher temperature ($+100$ °C, toluene- d_8), these pairs of separated signals coalesced into a set of single sharp signals. These spectral phenomena suggested that some or two conformers of the ten-membered ring of **1** are rapidly inverting each other at $+100$ °C and they are separated into two conformers or two groups of conformers in NMR time scale at below -28 °C. Then, empirical force field calculations were carried out in order to obtain conformational information. The ten-membered carbocycle is so small that two endocyclic double bonds must be placed perpendicularly to an averaged plane of the ten-membered ring. Therefore, the combination of two factors determining the conformation of **1**, direction of two double bond planes and a dihedral angle $\text{C}(6)\text{-C}(7)\text{-C}(8)\text{-C}(9)$, claimed eight strain minimum conformations. They were $\alpha\alpha\text{CT}$, $\alpha\alpha\text{CC}$, $\alpha\beta\text{TC}$, $\alpha\beta\text{TT}$, $\beta\beta\text{CT}$, $\beta\beta\text{CC}$, $\beta\alpha\text{TC}$, and $\beta\alpha\text{TT}$ -**1a** which were depicted in Fig. 2 with assistance of a molecular model. For the calculations a germacrene-type compound **5** was adopted as an adequate model of **1** since a long side chain of **1** effected apparently little on conformational stability of the ten-membered ring. Results of the calculations⁵⁾ performed with each of the eight principal conformers showed that the two conformers $\alpha\alpha\text{CT}$ and $\alpha\beta\text{TT}$ were equilibrated at room temperature in a ratio 60:40 (Table 1). We concluded that the two sets of signals exhibited in cmr were due to

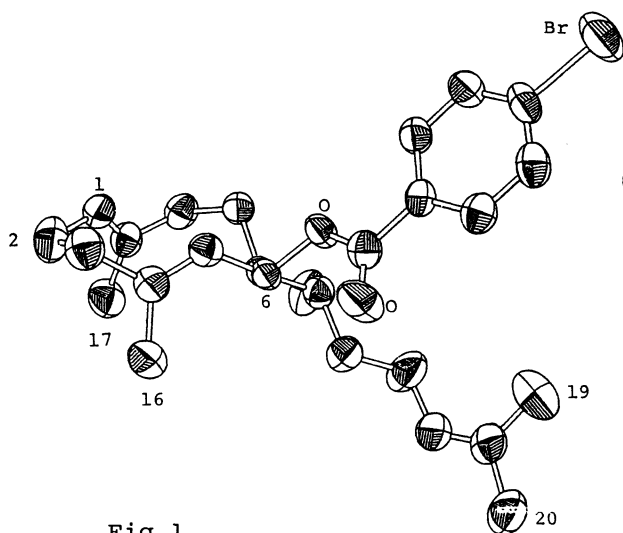


Fig. 1.

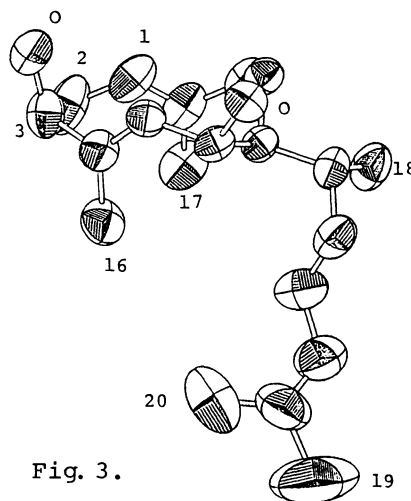


Fig. 3.

Fig. 2.

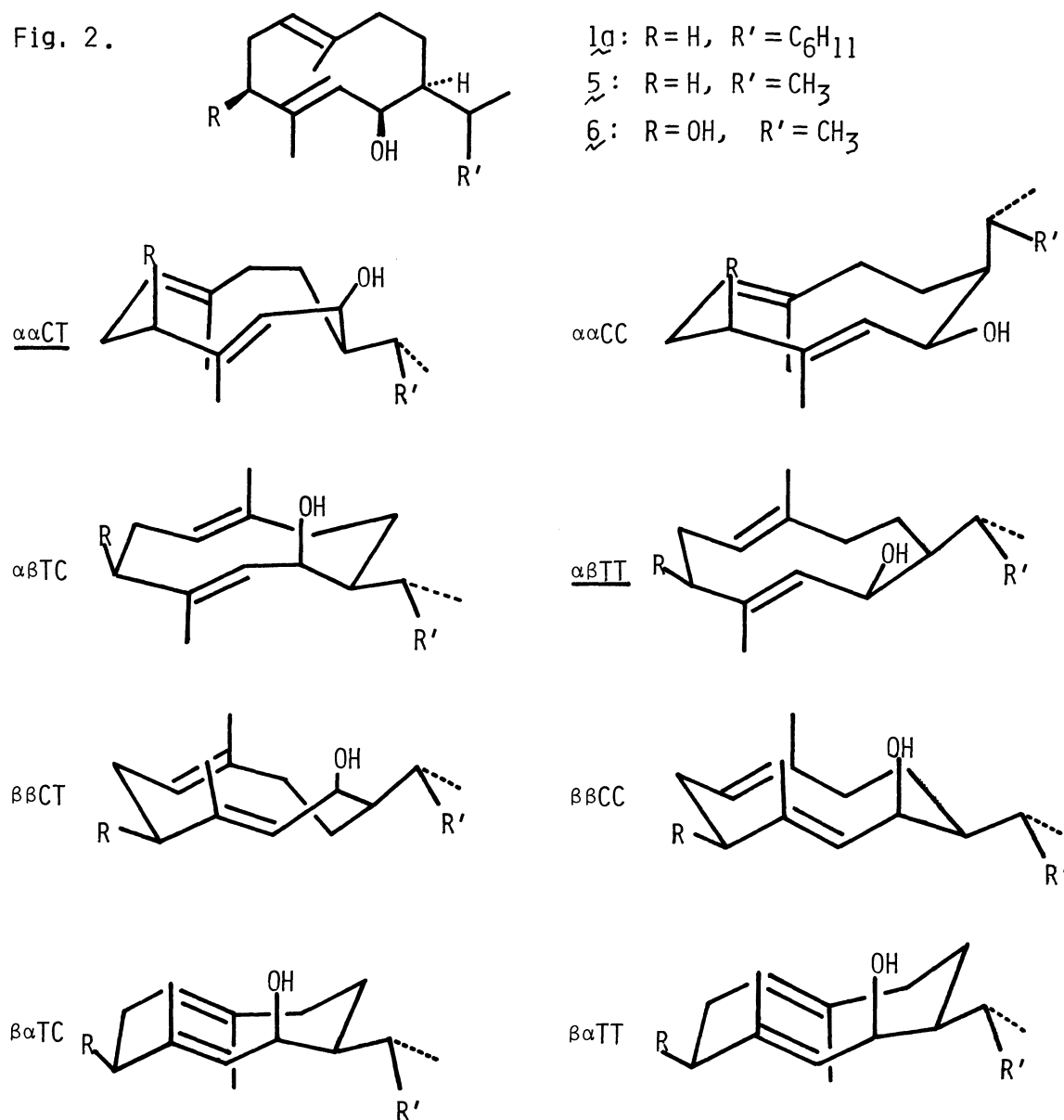


Table 1. Relative strain energy (kJ/mol) of eight strain minimum conformers. (Populations are also shown in parenthesis)

	$\alpha\alpha$ CT	$\alpha\alpha$ CC	$\alpha\beta$ TC	$\alpha\beta$ TT
5	0.00 (59.8%)	8.91 (1.6%)	22.84 (0.0%)	1.09 (38.6%)
6	0.00 (62.0%)	10.46 (0.9%)	24.64 (0.0%)	1.34 (36.0%)
	$\beta\beta$ CT	$\beta\beta$ CC	$\beta\alpha$ TC	$\beta\alpha$ TT
5	41.17 (0.0%)	21.76 (0.0%)	21.92 (0.0%)	49.12 (0.0%)
6	35.65 (0.0%)	13.64 (0.0%)	13.93 (0.0%)	40.58 (0.0%)

the above two conformers and the results of the calculations had been realized by NMR spectra. The most popular conformer α CT appeared in the crystal form shown by X-ray analysis.

A new germacrene-type diterpene, 3-acetoxyacetyldilophol (3), (mp 73-74 °C, $[\alpha]_D^{25} +7.9^\circ$ (c 2.2, CHCl₃)), was crystallized from hexane. The crystalline material had molecular formula C₂₄H₃₈O₄ (m/z M⁺ found 390.2790, calcd 390.2772), which together with IR bands at 1745 and 1240 cm⁻¹, suggested that the compound was a diacetate of a diterpene diol. This was supported by the facts that treatment of acetate 3 with lithium aluminum hydride in ether gave the corresponding diol 4, (C₂₀H₃₄O₂, mp 105-106 °C) and the mass spectrum of 3 showed peaks at 330 (M⁺-AcOH) and 270 (M⁺-2AcOH). The 400 MHz ¹H-NMR spectrum of 3 exhibited broad signals at δ 0.97 (3H, W_H=22 Hz, 18Me), 1.52 (3H, brs), 1.57 (3H, brs), 1.64 (3H, bs), 1.68 (3H, brs), 2.03 (3H, brs), 2.09 (3H, brs), 5.05 (1H, t, J=7 Hz, H on a trisubstituted double bond in the side chain), 5.08-5.32 (3H, complex pattern, two ten-membered olefinic protons and one acetoxymethine proton), 5.69 (1H, brd, J=7 Hz, an acetoxymethine proton). In order to establish the structure of 4, X-ray crystallographic study was undertaken. The crystal data for 4 were as follows: orthorhombic, space group P2₁2₁2₁, a=20.933(9), b=22.577(6), c=16.687(4) Å, Z=16, D_C=1.032 g cm⁻³. Intensities data of 5896 unique reflections for 2 θ <130° were obtained in the same manner as used for the preceding measurements. The structure was elucidated by the Monte Carlo direct method⁴); the 819th random phase set for the 70 strongest reflections led to the correct solution. Block-diagonal least-squares refinements including 108 hydrogen atoms reduced the R value to 0.085. One of the skeletons of the four independent molecules is illustrated in Fig. 3 which shows absolute configuration. Concerning ten-membered ring, 3 takes almost same conformation as 1 in crystalline state. The ¹³C-NMR spectrum (CDCl₃, room temperature) of 3-acetoxyacetyldilophol showed many broad and small peaks, together with sharp and strong signals (δ 17.1, 17.6, 25.2, 25.7, 30.3, and 34.1) due to the side chain. At temperature lower than -28 °C (toluene-d₈) two sets of sharp signals with an intensity ratio of about 3:2 were observed at δ 76.6 and 75.3. Very similar spectral feature to the preceding case prompted us to do again force field calculations. The calculations concerning a model compound 6 with an abbreviated side chain gave also very similar results to the case of 1 (Table 1). The diacetate 3 was considered to be equilibrated between two conformers, α CT and α TT in a ratio 60:40 at room temperature.

References

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